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20-21 October, Bor Lake, Serbia

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TABLE OF CONTENTS

1.	Invited lecture: Yuhui Zhang, Shuhong Liu, Yuling Liu; Mentor: Yong Du (China)	
	MICROSTRUCTURAL SIMULATION OF AGEING PRECIPITATION BASED ON	1
	THE DIFFUSION STUDY OF THE HCP_A3 PHASE IN Mg-Al-Sn ALLOYS	
2.	Student: Marina Marković; Mentor: Milan Gorgievski (Serbia)	
	REMOVAL OF COPPER IONS FROM AQUEOUS SOLUTIONS USING ONION	2
	PEELS AS AN ADSORBENT	
3.	Students: Nizama Baručija, Armin Čaušević, Merjem Delibašić; Mentor: Hasan	
	Avdušinović (Bosnia and Herzegovina) INFLUENCE OF GRAPHITE MORPHOLOGY ON THERMAL CONDUCTIVITY	2
		3
4.	Student: Alexandr Chesnyak; Mentor: Tamara Tikhomirova (Russia)	
	WAYS TO SOLVE ALTERNATIVE ENERGY SOURCES	4
5.	Student: Nikolay Palienko; Mentor: Tamara Tikhomirova (Russia)	
	DEVELOPMENT OF GEOTHERMAL ENERGY IN THE WORLD	7
6.	Student: Andrey Slyunkin; Mentor: Tamara Tikhomirova (Russia)	
	THE USE OF BIOENERGY RESOURCES IN THE PRODUCTION OF ELECTRICITY	10
7.	Students: Alida Kusić, Ilma Bošnjak; Mentor: Miliša Todorović (Bosnia and	
	Herzegovina)	
	SAFETY AND HEALTH IN COKING PLANTS THROUGH THE APPLICATION OF	13
	ENGINEERING MEASURES	
8.	Student: Aleksandra Radić; Mentor: Danijela Voza (Serbia)	
	METHODS FOR PRIORITISATION OF SUSTAINABLE DEVELOPMENT GOALS	14
9.	(SDGS) - AN OVERVIEW Student: Marija Kovač; Mentor: Snežana Vučetić (Serbia)	
9.	NON-DESTRUCTIVE TESTING OF INORGANIC MATERIALS AS	17
	DECISION TOOL IN CULTURAL HERITAGE	17
10		
10.	Student: Edita Bjelić; Mentors: Mersiha Suljkanović, Jasmin Suljagić (Bosnia and	
	Herzegovina) HYDROPHOBIC DEEP EUTECTIC SOLVENTS: PROMISING GREEN MEDIA FOR	18
	BIOMASS TREATMENT	10
11.	Student: Miloš Vuleta; Mentor: Jasmina Petrović (Serbia)	
	CONSIDERATION OF THE INFLUENCE OF STIR CASTING PROCESS	19
	PARAMETERS ON OBTAINING MMC CASTINGS	
12.	Students: Nizama Baručija, Resul Čehajić, Mahir Dreco; Mentors: Almaida Gigović-	
	Gekić, Amna Hodžić (Bosnia and Herzegovina)	
	INFLUENCE OF MIXING OF QUENCHING MEDIA ON MICROSTRUCTURE AND	20
12	HARDNESS OF STEEL 23MnB4	
13.	Students: Mahir Dreco, Armin Čaušević; Mentors: Branka Muminović, Behar Alić, Almaida Gigović-Gekić (Bosnia and Herzegovina)	
	TESTING OF WELDED JOINTS WITH LIQUID PENETRANTS	21
14.	Students: Vedran Milanković, Tamara Tasić; Mentor: Tamara Lazarević-Pašti	4 1
14.	(Serbia)	
	REMOVAL OF CHLORPYRIFOS AND MALATHION USING SPENT COFFEE	22
	GROUNDS – ISOTHERM STUDY	



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REMOVAL OF CHLORPYRIFOS AND MALATHION USING SPENT COFFEE GROUNDS – ISOTHERM STUDY

Students: Vedran Milanković, Tamara Tasić

Mentor: Tamara Lazarević-Pašti

University of Belgrade, VINČA Institute of Nuclear Sciences, Belgrade, Serbia

Abstract

In recent years, the extensive use of chlorpyrifos (CHP) and malathion (MLT), two common organophosphate (OP) insecticides, has raised concerns about their adverse effects on the environment and human health [1]. These pesticides can contaminate water bodies, soil, and food, posing potential risks to non-target organisms and human populations [2, 3]. Therefore, there is a growing demand for effective and sustainable methods to remove these pesticides from the environment [4]. Spent coffee grounds (SCG), a readily available agricultural waste, have shown promising potential as an adsorbent [3]. The aim of this study was to investigate the removal of CHP and MLT using SCG as an adsorbent and to examine the adsorption behavior through isotherm analysis.

Isotherm analysis was performed using four isotherm models: Freundlich, Langmuir, Temkin, and Dubinin-Radushkevich. The experimental data best fit the Langmuir isotherm model, suggesting a monolayer adsorption process on homogeneous adsorption sites. According to the Langmuir isotherm, the maximum adsorption capacity of SCG for CHP and MLT is 2.34 mg/g and 7.04 mg/g, respectively. Additionally, the Freundlich isotherm model fitted the experimental data for MLT adsorption on SCG very well, implying multilayer physisorption even after all adsorption sites are occupied. These findings provide valuable insights into the feasibility of SCG as an eco-friendly approach for the removal of CHP and MLT from the environment. The energy of adsorption obtained from the Dubinin-Radushkevich isotherm confirmed that in the case of CHP adsorption the binding is stronger than in the case of MLT adsorption.

Keywords: Organophosphorous pesticides, Adsorption, Biowaste, Spent Coffee Grounds, Isotherms

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